

# Process Hazards Associated with Carbon Capture and Sequestration



# Carbon Dioxide

## ➤ Carbon Dioxide

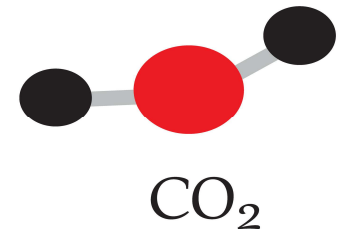
- Accounts for approximately 65% of greenhouse gas emissions in the US
- Approximately 15% of the CO<sub>2</sub> emitted is from oil & gas operations/facilities (amine vent stream or engine exhaust)
- Various rules, regulations and initiatives are in place that make it worth-while to reduce these emissions



# Carbon Dioxide

## ➤ Carbon Dioxide

- Naturally occurring portion of the atmosphere = 0.04%
- Found in the gas state at room temperature
- At normally-encountered concentrations it is odorless
- Carbon dioxide is 53% more dense than dry air but is long lived and thoroughly mixes in the atmosphere
- Carbon dioxide is soluble in water, in which it forms  $\text{H}_2\text{CO}_3$  (carbonic acid), which is a weak acid
- Atmospheric  $\text{CO}_2$  is the primary carbon source for life on Earth.
- In the air, carbon dioxide is transparent to visible light but absorbs infrared radiation, acting as a greenhouse gas



# Carbon Capture

- Most industrial CO<sub>2</sub> emissions are the result of combustion of fossil fuels
- A major sources of CO<sub>2</sub> in the oil & gas industry are from fuel-fired equipment used in the production, treatment and transportation of oil & gas
  - Heater/treaters at the wellhead
  - Product pumps and gas compressors
  - Heat medium equipment at oil & gas treatment/processing facilitates
  - Amine treatment process\*



# Carbon Capture

- Carbon capture, utilization and sequestration isn't new to the oil and gas industry
  - CO<sub>2</sub> has been separated from the production stream and used for enhanced oil recovery (EOR) since the 1970's
- Previous CO<sub>2</sub> inventories did not account for CO<sub>2</sub> generated from combustion sources
  - Companies can utilize low pressure capture, such as a vapor recovery unit to minimize release of combustion products to the atmosphere
  - These emissions can then be processed to separate gas components for reinjection



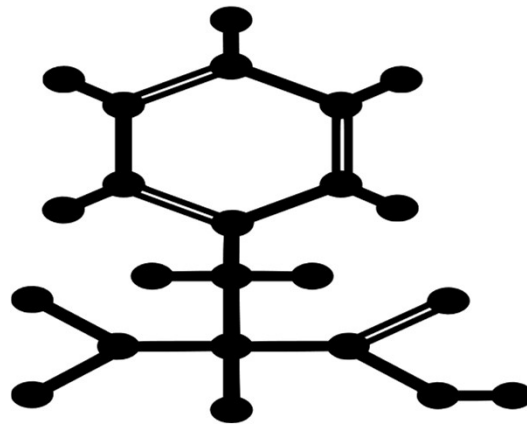
# Carbon Capture

- CO<sub>2</sub> exhaust streams from fuel-fired equipment is typically high-temperature (600°F to 900°F), low-pressure (approximately 15 psig)
  - Higher temperature piping and equipment requirements than normal production, treatment, or compression
  - Increased worker heat stress around equipment
  - Oxygen evacuation around leaks



# Carbon Capture

- CO<sub>2</sub> exhaust streams from fuel-fired equipment also contain a high level of impurities; such as oxygen and nitrogen compounds, and particulates
  - The higher level of impurities in the CO<sub>2</sub> stream requires lower temperature and higher pressure to transport efficiently



# Carbon Capture

- High-temperature CO<sub>2</sub> needs to be cooled to effectively transport
  - Typically used for cross exchange in the treatment process
  - Can be used for onsite power generation
  - Worse case is release of heat to atmosphere
- Low-pressure CO<sub>2</sub> needs to be compressed to effectively transport
- Impurities in the CO<sub>2</sub> stream need to be removed
  - Typically done using chemical removal (catalytic reduction or absorption)





# Carbon Dioxide Transportation

- CO<sub>2</sub> transportation
  - CO<sub>2</sub> can be transported in many ways, but using a pipeline is the most efficient
  - CO<sub>2</sub> can be transported in a gaseous state or in a supercritical state, with transport in a supercritical state preferred for longer distance transport due to higher volume capability and lower loss during transport
  - To get CO<sub>2</sub> to a supercritical state it must be kept at 88°F or lower and 1,010 psig or higher
    - With the higher level of impurities, the temperature and pressure requirements change (temperature goes lower, pressure goes higher)



# Carbon Dioxide Transportation

## ➤ CO<sub>2</sub> transportation

- A CO<sub>2</sub> pipeline is regulated by DOT PHMSA as a Hazardous Liquids Pipeline under 49CFR195 (>4,500 miles currently in existence)
  - Need to identify High Consequence Area
  - Decreased, automatic valve spacing
  - Increased inspections/surveys



# Carbon Dioxide Transportation

## ➤ CO<sub>2</sub> transportation

- Transport in a supercritical state requires components of the system must be chosen for the temperature and pressure environment in which the components will be used so that the pipeline will maintain its structural integrity
  - NOT typical Carbon Steel, however, can be used with appropriate ceramic/polymer coatings
  - 316 Stainless Steel (Cr, Ni, Mo) preferred
  - Polymer piping (pressure limitations)
- Pipe joint welds must also be appropriate materials
- May cause problems when converting existing systems



# Carbon Dioxide Transportation

- CO<sub>2</sub> transportation
  - Increased transport distance also requires additional equipment and treatment to maintain the supercritical state (compressors, coolers and treatment skids)



# Carbon Sequestration

- CO<sub>2</sub> injection and storage
  - Reservoir availability and compatibility
    - Depleted oil & gas reservoirs are perfect for sequestration
    - Regulated as Class VI injection wells under the Safe Drinking Water Act by USEPA and some states (LA, ND, WV, WY)
    - Requirements in place for plume impact analysis, well design, and operating procedures
    - Currently 11 EPA and 30 state permitted Class VI wells (>300 permit aps)



# Carbon Sequestration

- CO<sub>2</sub> injection and storage
  - CO<sub>2</sub> can be used as a fracturing media.
    - CO<sub>2</sub> interacts with reservoir rock under higher pressure conditions, effectively changing the pore structure of the reservoir by dissolution of the rock matrix, and increasing the number and complexity of fractures
    - CO<sub>2</sub> has a higher solubility in formation fluid under high-pressure conditions and diffuses more deeply into the formation



# Carbon Sequestration

- Alternative sequestration methodology
  - In Oman and Iceland collected  $\text{CO}_2$  is being injected into peridotite and volcanic formations, a chemical reaction converts the  $\text{CO}_2$  into solid carbonates



# Carbon Sequestration

- The Bipartisan Budget Act of 2018 established a tax credit for carbon sequestration (Section 45Q)
  - Amount of credit is based upon date sequestration started and how it is accomplished (extended under the BBB)
  - Can range from \$10/metric ton captured and sold to \$85/metric ton sequestered/used
- Carbon offsetting:
  - Investing in carbon negative projects (Currently - EPA grant program, PA, CA; several other states considering)
  - Purchasing carbon offset credits





# Thank You!

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